

ENERGY PRODUCTION FROM WASTE WATER USING MFC TECHNOLOGY

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ABSTRACT

With the escalating global energy demand and the pressing need for sustainable wastewater treatment solutions, microbial fuel cell (MFC) technology emerges as a promising innovation to address both challenges simultaneously. This project report delves into the intricate realm of energy production from wastewater through MFCs, exploring their potential as an eco-friendly and efficient approach. The foundation of this research lies in comprehending the fundamental principles of MFC technology, where microbial communities catalyse the oxidation of organic matter in wastewater, resulting in electrical energy generation. The report elucidates the intricate interplay of microorganisms, electrodes, and electron transfer mechanisms, providing a comprehensive understanding of MFC functioning. The study takes a multidimensional approach, investigating the integration of wastewater treatment and energy generation. By analysing existing literature, this report highlights the significant advancements made in MFC technology, emphasizing their remarkable capacity to treat wastewater while concurrently producing renewable energy.

KEYWORDS: sustainable wastewater treatment, promising innovation, MFC, eco-friendly, significant advancements, renewable energy

1. Introduction

The increasing global demand for energy and the growing concerns about environmental pollution have compelled researchers to explore innovative and sustainable approaches to tackle these challenges. One such area of interest is the utilization of wastewater, a substantial and often underutilized resource, as a potential source of renewable energy while simultaneously addressing the critical issue of wastewater treatment. Wastewater, generated from domestic, industrial, and agricultural activities, contains a significant amount of organic matter and other pollutants. Traditional wastewater treatment methods, such as activated sludge processes and anaerobic digestion, have been employed to remove pollutants and reduce environmental harm. However, these methods are energy-intensive and may not fully harness the latent energy present in wastewater. Microbial Fuel Cell (MFC) technology, a revolutionary concept that gained momentum in the early 21st century, offers a promising solution to this conundrum. The core principle of MFCs involves using



naturally occurring microorganisms to catalyse the electrochemical reaction between organic matter in wastewater and an electrode surface, resulting in the generation of electrical energy. Simultaneously, this process aids in the breakdown of organic pollutants, effectively treating the wastewater. The concept of MFCs dates back to the early 20th century, but it was not until the late 20th and early 21st century that significant advancements in microbial research and electrochemistry propelled MFCs into the spotlight. Researchers worldwide began exploring various configurations, electrode materials, and microbial consortia to enhance MFC performance and practical applications. MFCs have emerged as a versatile technology with dual benefits: generating electricity from wastewater, a renewable and readily available resource, and providing a more energy-efficient and environmentally friendly approach to wastewater treatment.

2.METHODOLOGY

DESIGN AND DESCRIPTION OF MFC

SAMPLE COLLECTION

MICROBIAL FUEL CELL ASSEMBLY

OPERATION OF MFC

MEASUREMENT OF THE POWER GENERATION

COMPARING THE POWER GENERATED



2.1 SELECTION OF ELECTRODE MATERIALS

Carbon cloth is a widely used and popular electrode material in Microbial Fuel Cells (MFCs) due to its excellent properties that enhance MFC performance. Carbon cloth electrodes offer several advantages that make them ideal for various applications in MFC technology:

- 1. High Conductivity: Carbon cloth is a highly conductive material, which facilitates efficient electron transfer between the microorganisms and the electrode surface. This high conductivity allows for enhanced electrical performance, resulting in improved power generation.
- 2. Large Surface Area: Carbon cloth has a porous and three-dimensional structure, providing a large surface area for microbial attachment and biofilm formation. The increased surface area allows more microorganisms to interact with the electrode, leading to higher current generation and improved MFC efficiency.
- 3. Biocompatibility: Carbon cloth is generally considered biocompatible, meaning it supports the growth and activity of electrochemically active bacteria. Its surface properties allow for strong adhesion of microbial biofilms, which are essential for effective electron transfer in MFCs.
- 4. Chemical Stability: Carbon cloth exhibits good chemical stability, which is crucial for withstanding the harsh conditions within MFCs. It can resist degradation and maintain its performance during prolonged operation.
- 5. Flexibility and Versatility: Carbon cloth is flexible and can be easily moulded or shaped to fit various MFC configurations, making it versatile for different electrode designs. This flexibility allows researchers to customize the electrode geometry to suit specific applications.
- 6. Ease of Handling: Carbon cloth is relatively easy to handle and assemble in MFC systems, simplifying the fabrication process compared to some other electrode materials.
- 7. Cost-effectiveness: Carbon cloth is generally more cost-effective compared to certain other highperformance materials, such as graphene or platinum. Its reasonable cost makes it attractive for largescale applications and research studies with budget constraints.
- 8. Long Service Life: Carbon cloth electrodes can have a long service life, particularly when properly maintained. This durability ensures stable MFC performance over extended periods.
- 9. Compatibility with Various Wastewater Types: Carbon cloth electrodes have been successfully employed in MFCs for various wastewater types, including domestic, industrial, and agricultural wastewaters. Their compatibility with different wastewater sources makes them suitable for a wide range of applications.

Despite its numerous advantages, carbon cloth also has some limitations. For instance, it may have slightly lower surface area compared to certain nanostructured materials like graphene. However, the overall performance of carbon cloth electrodes remains excellent, making them a popular choice for researchers and practitioners working in the field of MFC technology.

2.2 ELECTRODE CONFIGURATION

Porous Carbon Cloth Configuration:

In this configuration, carbon cloth with a porous structure is used to increase the effective surface area available for microbial colonization. The porous structure allows microorganisms to penetrate into the carbon cloth, enhancing their interaction with the electrode surface.



The specific choice of carbon cloth electrode configuration depends on the intended application, desired performance, and design considerations. Factors such as electrode surface area, electron transfer efficiency, and ease of assembly play a role in determining the most suitable configuration for a particular MFC system. Researchers continuously explore innovative electrode configurations to optimize MFC performance and explore new applications in sustainable energy generation and wastewater treatment.



Single Stainless Steel Mesh Configuration:

In the single stainless steel mesh configuration, a single piece of stainless steel mesh serves as both the anode and cathode in the MFC. The mesh is typically coated with a conductive material, such as carbon or a conductive polymer, to enhance electron transfer. The anode and cathode regions of the mesh are separated by a proton exchange membrane (PEM) or anion exchange membrane (AEM) to prevent direct contact between them while allowing ion transport.





Packaged flexible solid-state supercapacitor

3.WASTEWATER CHARACTERISATION

- 1. Food waste
- 2. Tannery wastewater

Food wastewater, also known as food processing wastewater or food industry effluent, is generated from various stages of food production, processing, and preparation. The characteristics of food wastewater can vary depending on the specific type of food industry, the processes involved, and the level of treatment applied. However, some common characteristics of food wastewater include:

- 1. High Organic Content: Food wastewater is typically rich in organic matter, including carbohydrates, proteins, lipids, and other organic compounds. This organic content makes food wastewater a potential source of energy for microbial processes, such as in anaerobic digestion or microbial fuel cells.
- 2. High Biochemical Oxygen Demand (BOD): The high organic content in food wastewater contributes to its elevated Biochemical Oxygen Demand (BOD). BOD is a measure of the amount of oxygen

required by microorganisms to biologically degrade the organic matter in the water. High BOD levels can lead to oxygen depletion in receiving water bodies, affecting aquatic life.

- 3. High Chemical Oxygen Demand (COD): Food wastewater also exhibits a high Chemical Oxygen Demand (COD), which represents the total quantity of oxygen required for both biological and chemical oxidation of organic matter. Elevated COD levels indicate a large amount of readily biodegradable organic compounds.
- 4. Nutrient Content: Food wastewater may contain significant amounts of nutrients, such as nitrogen and phosphorus, which are present in the form of proteins and other organic compounds. These nutrients can contribute to eutrophication if not adequately removed during wastewater treatment.
- 5. Suspended Solids: Food wastewater can contain suspended solids, including food particles, fats, and other materials used in food processing. High levels of suspended solids can lead to turbidity and potential environmental issues.
- 6. Fats, Oils, and Grease (FOG): Food wastewater from certain food industries, such as restaurants and fast-food chains, can contain substantial amounts of fats, oils, and grease (FOG). FOG can cause blockages in sewer systems if not properly managed.
- 7. pH Variability: The pH of food wastewater can vary widely depending on the type of food industry and the specific processes involved. Some food processing activities may produce acidic or alkaline wastewater, which can require pH adjustment during treatment.
- 8. Colour and Odour: Certain food wastewater streams may exhibit distinct colours and odours, particularly from food additives, spices, and food processing chemicals.
- 9. Seasonal Variability: Food wastewater characteristics may show seasonal variations due to changes in production levels, food types processed, and ingredient availability.
- 10. Pathogens and Contaminants: Food wastewater may also carry pathogenic microorganisms and chemical contaminants if not properly treated, posing health and environmental risks.

Tannery wastewater is generated from the leather tanning industry, and it is known for its high pollutant load and complex composition. The characteristics of tannery wastewater can vary depending on the tanning processes, raw materials used, and treatment practices. Here are some common characteristics of tannery wastewater:

- High Organic Content: Tannery wastewater is characterized by a high organic content, primarily from organic matter such as proteins, fats, and complex tanning agents used in the leather-making process. This high organic load contributes to the elevated Biochemical Oxygen Demand (BOD) and Chemical Oxygen Demand (COD) of tannery wastewater.
- 2. High Total Dissolved Solids (TDS): Tannery wastewater often contains significant amounts of dissolved salts, such as sodium chloride and other chemicals used during the tanning process. The high TDS levels can lead to salinity issues in receiving water bodies if not adequately treated.
- 3. Chromium and Heavy Metals: One of the most significant concerns in tannery wastewater is the presence of chromium and other heavy metals. Chromium is used in the tanning process to stabilize leather and impart specific properties. However, if not properly treated, it can be highly toxic to aquatic life and pose risks to human health.



- 4. Sulphides: Tannery wastewater can contain sulphides originating from chemicals used in the deliming and liming steps of leather processing. Sulphides contribute to the characteristic odour associated with tannery wastewater.
- 5. High pH: Tanning processes often involve the use of alkaline chemicals, which can result in high pH levels in the wastewater. The pH of tannery wastewater is typically in the alkaline range.
- 6. Color and Turbidity: Tannery wastewater may exhibit a dark color and high turbidity due to the presence of tannins and other organic compounds.
- 7. Chlorinated Organic Compounds: Some tanneries may use chlorinated chemicals as biocides, which can lead to the presence of chlorinated organic compounds in the wastewater.
- 8. High Temperature: Tannery wastewater is often warm due to the use of hot water in various processes, which can have thermal impacts on receiving water bodies if not managed properly.
- 9. High Nutrient Content: Tannery wastewater may contain elevated levels of nutrients, such as nitrogen and phosphorus, from the use of fertilizers and other additives in the leather-making process.
- 10. Pathogens: Untreated tannery wastewater can harbour pathogenic microorganisms, presenting risks to public health and the environment if discharged without proper treatment.

4.RESULTS









It is found that the MFC setup with tannery effluent produces power higher than the food waste and combined waste (food and tannery).

5.CONCLUSION

Microbial Fuel Cell (MFC) technology holds great promise as a sustainable and innovative approach for energy production and wastewater treatment. Over the years, extensive research and development have advanced our understanding of MFCs, paving the way for their potential real-world applications. The characteristics of MFCs, such as their ability to generate electricity from organic matter while simultaneously treating wastewater, make them attractive for a range of environmental and energy-related challenges. Microbial Fuel Cells represent a remarkable convergence of microbiology, electrochemistry, and environmental engineering. Their ability to harness the power of microorganisms for sustainable energy generation and wastewater treatment holds the potential to make a meaningful impact on our journey toward a more sustainable and greener future. By continuing to explore new innovations, optimizing performance, and overcoming technical hurdles, MFCs can emerge as a transformative technology in the realm of renewable energy and environmental protection.

REFERENCE

- 1) Rabaey, K., Verstraete, W. (2005). Microbial fuel cells: novel biotechnology for energy generation. Trends in Biotechnology, 23(6), 291-298.
- 2) Logan, B. E., & Regan, J. M. (2006). Electricity-producing bacterial communities in microbial fuel cells. Trends in Microbiology, 14(12), 512-518.
- Liu, H., Ramnarayanan, R., & Logan, B. E. (2004). Production of electricity during wastewater treatment using a single chamber microbial fuel cell. Environmental Science & Technology, 38(7), 2281-2285.
- 4) Kim, J. R., Dec, J., & Logan, B. E. (2008). Removal of odors from swine wastewater by using microbial fuel cells. Applied Microbiology and Biotechnology, 78(3), 451-457.
- Zhang, F., Zhang, Y., Quan, X., & Chen, S. (2009). Enhanced electricity generation in microbial fuel cells with high concentration of formate. Bioresource Technology, 100(2), 717-723.
- 6) Aelterman, P., Rabaey, K., Pham, H. T., Boon, N., Verstraete, W. (2006). Continuous electricity generation at high voltages and currents using stacked microbial fuel cells. Environmental Science & Technology, 40(10), 3388-3394.
- 7) Zhao, F., Harnisch, F., Schroder, U., Scholz, F., Bogdanoff, P., Herrmann, I. (2006). Application of pyrolysed iron (II) phthalocyanine and CoTMPP based oxygen reduction catalysts as cathode materials in microbial fuel cells. Electrochemistry Communications, 8(5), 869-873.
- 8) Kaur, A., Liu, H., & Sinton, D. (2018). Advances in Microbial Fuel Cell (MFC) architectures and materials: A review. Bioresource Technology, 258, 353-365.
- 9) Kim, J. R., Min, B., & Logan, B. E. (2005). Evaluation of procedures to acclimate a microbial fuel cell for electricity production. Applied Microbiology and Biotechnology, 68(1), 23-30.
- 10) Logan, B. E. (2009). Exoelectrogenic bacteria that power microbial fuel cells. Nature Reviews Microbiology, 7(5), 375-381.
- 11) Logan, B.E. (2008). Microbial Fuel Cells. John Wiley & Sons, Inc.

- 12) Rabaey, K. and Rozendal, R.A. (2010). Microbial electrosynthesis revisiting the electrical route for microbial production. Nature Reviews Microbiology, 8, 706-716.
- 13) Kim, J.R., Premier, G.C., Hawkes, F.R., and Dinsdale, R.M. (2006). A Review of the Applications of Microbial Fuel Cells. Bioresource Technology, 98(11), 2309-2315.
- 14) Liu, H. and Logan, B.E. (2004). Electricity Generation Using an Air-Cathode Single Chamber Microbial Fuel Cell in the Presence and Absence of a Proton Exchange Membrane. Environmental Science & Technology, 38(14), 4040-4046.
- 15) Lovley, D.R. (2006). Microbial Fuel Cells: Novel Microbial Physiologies and Engineering Approaches. Current Opinion in Biotechnology, 17(3), 327-332.
- 16) Pham, T.H., Rabaey, K., Aelterman, P., Clauwaert, P., and Verstraete, W. (2006). Microbial Fuel Cells in Relation to Microbial Ecology. Applied Microbiology and Biotechnology, 72(4), 589-597.
- 17) Sleutels, T.H.J.A., Darus, L., and Hamelers, H.V.M. (2011). Advancements and Perspectives of Microbial Fuel Cells. Renewable and Sustainable Energy Reviews, 15(1), 1-8.
- 18) Zhang, F., Cheng, S., Pant, D., and Van Bogaert, G. (2017). Recent Progresses in Microbial Fuel Cells: A Review. Journal of Renewable and Sustainable Energy Reviews, 68, 545-555.
- 19) Zhou, M., Chi, M., Luo, J., He, H., Jin, T., and Lu, H. (2014). Microbial Fuel Cells: Advances, Applications, and Challenges. BioMed Research International, 2014, 1-20.
- 20) Logan, B.E., Wallack, M.J., Kim, K.Y., He, W., and Feng, Y. (2005). A Hybrid Microbial Fuel Cell (MFC)-Membrane Bioreactor (MBR) System for Enhanced Wastewater Treatment. Biotechnology and Bioengineering, 91(4), 491-497.
- 21) Aelterman, P., Rabaey, K., Pham, T.H., Boon, N., Verstraete, W., and De Schamphelaire, L. (2006). Continuous Electricity Generation at High Voltages and Currents Using Stacked Microbial Fuel Cells. Environmental Science & Technology, 40(10), 3388-3394.
- 22) Ishii, S., Shimoyama, T., Hotta, Y., and Watanabe, K. (2008). Characterization of a Sulfate-Reducing Bacterium That Rapidly Mineralizes Toluene in a Microbial Fuel Cell. Applied and Environmental Microbiology, 74(9), 2766-2768.
- 23) Min, B., Kim, J.R., Oh, S.E., and Regan, J.M. (2008). Electricity Generation from Swine Wastewater Using Microbial Fuel Cells. Water Research, 42(16), 4093-4098.
- 24) Rozendal, R.A., Hamelers, H.V.M., Rabaey, K., Keller, J., and Buisman, C.J.N. (2008). Towards Practical Implementation of Bioelectrochemical Wastewater Treatment. Trends in Biotechnology, 26(8), 450-459.
- 25) Santoro, C., Babanova, S., and Artyushkova, K. (2017). Recent Advances and Challenges of Fuel Cells and Microbial Fuel Cells. Current Opinion in Electrochemistry, 4(1), 123-128.
- 26) Wang, X., Feng, Y., and Ren, N. (2015). Carbon Felt-Based Electrodes for Microbial Fuel Cells: A Review. Journal of Power Sources, 279, 697-706.
- 27) Yen, H.W., Hu, C.C., Cheng, Y.S., and Chang, Y.F. (2015). Integration of Microbial Fuel Cell and Membrane Bioreactor for Electricity Generation and Simultaneous Wastewater Treatment: A Review. Journal of Chemical Technology and Biotechnology, 90(5), 775-783.

