

Material of Aluminium-Fly Ash

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Abstract - In comparison to unmodified alloys, metal matrix composites (MMCs) exhibit significantly improved properties such as high specific strength, specific modulus, soaking capacity, and good wear resistance. Composites with low density and low cost reinforcement are gaining increased interest. Among the numerous dissenters used, ash is one of the most affordable and low-density reinforcements available in large quantities as a solid waste product of coal combustion in thermal power plants. As a result, composites reinforced with ash are likely to be prohibitively expensive for widespread use in automotive and small engine applications. As a result, it is expected that the incorporation of ash particles into aluminium alloys will encourage another use for this low-cost waste while also conserving energy intensive aluminium and thus lowering the cost of aluminium products. Nowadays, particulate reinforced aluminium matrix composites are gaining popularity due to their isotropic properties and the ability to undergo secondary processing to facilitate the fabrication of secondary components. The current investigation focuses on the beneficial use of abundant industrial waste fly ash by dispersing it in aluminium to produce composites via the stir casting method.

Key Words: composites, industrial waste, applied load and sliding velocity.

1. INTRODUCTION

Traditional monolithic materials are limited in their ability to combine strength, hardness, and density in an honest manner. Composites are the most promising material of recent interest for addressing these shortcomings and meeting the growing demand for contemporary technology. In comparison to unmodified alloys, metal matrix composites (MMCs) exhibit significantly improved properties such as high specific strength, specific modulus, soaking capacity, and good wear resistance. Composites with low density and low-cost reinforcement are gaining increased interest. Among the numerous dissenters used, ash is one of the most affordable and low-density reinforcements available in large quantities as a solid waste product of coal combustion in thermal power plants. As a result, composites reinforced with ash are likely to be prohibitively expensive for widespread use in automotive and small engine applications. As a result, it is expected that the incorporation of ash particles into aluminium alloys will encourage another use for this low-cost waste while also allowing for the conservation of energy-intensive aluminium, lowering the cost of aluminium products. Nowadays, particulate reinforced aluminium matrix composites are gaining popularity due to their isotropic properties and the ability to undergo secondary

processing to facilitate the fabrication of secondary components. In comparison to unreacted alloys, cast aluminium matrix particle reinforced composites have a higher specific strength, specific modulus, and good wear resistance. R.Q. Guo and P.K. Rohatgi investigated the prospect of employing fly-ash as a reinforcing material in aluminium melts and discovered that the electrical resistivity, low thermal conductivity, and low density of fly-high ash can be beneficial for the construction of lightweight insulating composites. By injecting reinforcing particles into the liquid matrix via the liquid metal route, a particulate composite can be prepared. The casting route is preferred because it enables low-volume production. Stir stirring is the simplest and least expensive method of producing liquid state. The only drawback to this process is the nonuniform distribution of annular material caused by low wetting capacity and gravity regulated separation. The size, shape, and volume fraction of aggregates, matrix materials, and the interface behaviour all affect the mechanical properties of composites. Numerous researchers have discussed these points. Rohatgi reports that as the volume percentage of fly ash in composites increases, the neural value increases in Al-fly ash (precipitator type) composites. Additionally, they reported that as the fly ash volume percentage increased, the ash alloy's volatility increased. Aghanian et al. reported significant improvement. Matrix and aggregated validation interfaces play a critical role in determining the properties of MMCs. Load transfer is necessary for strong and robust interfaces. The roughness of the interface is determined by the deflection of the crack, and the transition is determined by the relaxation of the top stress near the interface. Literacy, However, data on the friction and wear properties of ash reinforced AMC are extremely scarce. According to Rohatgi, adding fly ash particles to the active alloy significantly increases its photoresistance to decomposition. They attributed the increase in aluminosilicate component wear resistance to fly ash particles. Fly-ash composed primarily of silica, refractory oxides such as silica alumina, and iron oxide are used as reinforcing phases in this work. A composite containing 10% fly-ash as the strong phase was created. Additionally, the commercially pure user was melted and poured. Then, fly-ash was analysed for particle size and chemical composition. The mechanical, physical, and wear properties of the material were compared to those of a commercially pure user.

Composite

Composite material is a material composed of two or more distinct phases (matrix phase and strong phase) and differs significantly from any component in bulk properties. Many common materials (metals, alloys, doped ceramics and polymers mixed with additives) also have a small amount of scattered phases in their composition, although they are not

considered as composite materials because their properties are similar to their base components (Physical) has the same property of steel as that of pure iron). The favorable properties of composite materials are high hardness and high strength, low density, high – temperature stability, high electrical and thermal conductivity, adjustable coefficient of thermal expansion, corrosion resistance, superior wear resistance, etc..

I. Matrix phase

1. Primary stage, a continuous character,
2. Usually more ductile and less rigid phase,
3. Holds the strong phase and shares a load with it.

II. Reinforcing phase

1. The second phase (or phase) is written in a discontinuous form in the matrix,
2. It is usually stronger than the matrix, so it is sometimes called the reinforcement phase.

Composites as engineering materials generally refer to materials with the following characteristics:

- These are made artificially (thus, except for natural materials such as wood).
- These include at least two different species with a well-defined interface.
- Their properties are influenced by the volume percentage of the components.
- At least one of these assets is not held by the individual component.

The overall performance depends on:

- a) Properties of Matrix and Reinforcement,
- b) Size and distribution of components,
- c) Size of the components,
- d) Nature of interface between components.

1) Classification of Composites

Composite materials are classified as:

1. Based on the matrix content,
2. Depending on the filler material

III. Stir casting method of manufacturing MMC

Liquid state fabrication of metal matrix composites involves the dispersed state in the molten matrix metal, followed by its solidification.

To provide high levels of mechanical properties of the composite, good interfacial bonding (wetting) must be achieved between the dispersed state and the liquid matrix.

Wetting correction can be achieved by coating dispersion phase particles (fibers). Proper coating not only reduces interfacial energy, but also prevents

chemical contact between the dispersed state and the matrix.

The simplest and most effective method of liquid state formation is Stir casting.

IV. Stir casting

Stir casting is a liquid phase method of manufacturing composite materials, in which a dispersed phase (ceramic particles, small fibers) is mixed with a molten matrix metal by means of mechanical stirring.

Liquid composite materials are then cast by conventional casting methods and can also be processed by traditional metal forming technologies.

Stir casting is characterized by the following characteristics:

3. The dispersed phase content is limited (usually no more than 30 vol%. %).

The distribution of phase scattered throughout the matrix is not completely homogeneous:

1. Local clusters of dispersed particles (fibers)
 2. Gravitational separation of the dispersed state can occur due to the difference in density of dispersed and matrix phase.
- Technology is relatively simple and low cost.

If in the semi-solid state, the distribution of the scattered phase can be improved.

The method is called re-casting using a metal composite material in a semi-solid state. The high viscosity of the semi-solid matrix material enables better mixing of the dispersion phase.

V. Strengthening mechanism of composites

The reinforcement mechanisms of composites vary with different types of reinforcing agent morphology such as fiber, particulate or diffuse type reinforcing elements.

VI. Structure of fiber structure of mixed structure

In this type of composite, the reinforcing phase carries the bulk of the load and transfers the load to the reinforcing phase by the mechanism of the matrix seam. The high strength of the reinforcing phase restricts the free elongation of the matrix, particularly in its vicinity, while the latter is free to elongate at some distance from the former.

This type of non-uniform deformation of the matrix causes a shear stress at the matrix reinforcement interface resulting in tensile stresses in the reinforcing phase. Thus the stress is transferred to the reinforcing phase. Fibers can be either continuous or closed in the matrix. In the former case the load is directly applied to the strong phase and the stress is constant over its entire length. In the case of unsaturated fibers, the stress in the fiber increases to a zero value at the end of the maximum value at the center and thus the developed average tensile strength is always lower than that of continuous fibers. For the same when fracture of the reinforcing phase, therefore the strength of the closed fiber reinforced composite increases with increasing fiber length and the continuous fiber reinforcement of artifacts. Also, the strength of the fiber reinforced composite will be maximum when the fiber is aligned in the direction of applied stress i.e., in the impaction state. So, the strength of such a mixture depends on the volume fraction of the reinforcing element present in the composite, which can be determined by the simple rule of mixing.

The expression of the composition of the organized composition

The dispersion strengthening method uses a fine dispersion of mixed second stage reinforcing chemicals in a soft ductile matrix. Strong particles stifle clutter movement and reinforce the matrix. The primary principle that is reiterated here is that the matrix is strengthened by constructing clutter loops around the dispersion particles. As a result, dislocations have a tough time moving about the particles. The degree of strengthening is determined by several variables, including the volume percent of the dispersion phase, the degree of dispersion, the size and shape of the dispersion phase, and the differential particle spacing. The matrix materials in such composites bear the majority of the load.

Strong Mechanisms of Partnership Composites

Because particle reinforced composites have a particle size greater than $1\mu\text{m}$, they are strengthened in two ways. The first approach uses particles to transport loads along the matrix material, while the second method uses an incompatible interface between the particles and the matrix. Thus, a considerable amount

of clutter is formed at the interface, which strengthens the material. The degree of reinforcement is proportional to the volume, distribution, size, and particle size of the particulate, among other factors (volume fraction).

Fly ash is a byproduct of coal burning. It is a by-product of the industrial process of recovering flue gas from coal-fired power plants. The components of fly ash produced vary considerably depending on the source and composition of the coal to be burned, but all fly ash contains significant levels of silica (silicon dioxide, SiO_2) (both amorphous and crystalline) and lime (calcium oxide, CaO). To which belongs. Fly ash is composed primarily of SiO_2 , Al_2O_3 , and Fe_2O_3 with tiny amounts of Mg , Ca , Na , and K . The majority of fly ash particles are spherical in shape and range in size from $1100\mu\text{m}$ to $100\mu\text{m}$ [22], with a specific surface area of between 250 and $600\text{ m}^2/\text{Kg}$. Fly ash has a specific gravity of 0.6–2.8 g/cc . Fly ash's physical qualities are mostly determined by the way coal is burned and the conditions under which it is burned. Class F fly ash is typically made from high-rank (high carbon content) coals such as anthracite and bituminous coals, whereas class C fly ash is made from low-rank coals. Precipitators and cenosphere are two forms of fly ash particles. Typically, solid spherical fly ash is referred to as precipitator fly ash, while hollow fly ash with a density less than 1.0 gcm^{-3} is referred to as cenosphere fly ash. The most prevalent variety of fly ash contains crystalline compounds such as quartz, mulit, and hematite, as well as glassy compounds such as silica glass and various oxides. Precipitators fly ash, which has a density of 2.0–2.5 gcm^{-3} , can enhance the hardness, strength, and wear resistance of certain matrix materials while decreasing their density. Due to its low density, which ranges between 0.4–0.7 gcm^{-3} when compared to metal densities, cenosphere fly ash can be used to synthesise ultra-light composite materials. Is in Matrices with a density of 1.6–2.0 gcm^{-3} [23]. Coal fly ash has a variety of applications [24], including cement additives, masonry blocks, concrete mixtures, as a component of lightweight alloys, as a flux, and as a concrete aggregate. Structural fillers are utilised in the construction of highways and runways. Materials used in the manufacture of roofing granules and grouting. Although fly ash is mostly used in the cement and

concrete industries, innovative new uses for fly ash are being pursued, like as the use of fly ash in the creation of MMC.

Fly ash is one of the residues generated in coal combustion. It is an industrial by-product recovered from the flue gas of coal-fired power plants. Depending on the source and makeup of the coal to be burned, the components of the fly ash produced vary greatly, but all fly ash contains substantial amounts of silica (silicon dioxide, SiO_2) (both amorphous and crystalline) and lime (calcium oxide, CaO). In general, fly ash has SiO_2 , Al_2O_3 , Fe_2O_3 as major components and Mg, Ca, Na, K etc. as minor components. Fly ash particles are mostly spherical in shape and range from $1100\mu m$ to $100\mu m$ [22] with a specific surface area, typically between 250 and $600 m^2 / kg$. The specific gravity of fly ash varies in the range 0.6–2.8 g/cc . The physical properties of fly ash mainly depend on the burning and burning conditions of coal. Class F fly ash is generally produced from high-rank (high carbon content) coals such as anthracite and bituminous coals, while class C fly ash is produced from low-grade coals. Fly ash particles are classified into two types, precipitators and cenospheres. Typically, solid spherical particles of fly ash are called precipitator fly ash and hollow particles of fly ash with density less than $1.0 gcm^{-3}$ are called cenosphere fly ash. A common type of fly ash is usually composed of crystalline compounds such as quartz, mullite and hematite, glassy compounds such as silica glass and other oxides. Precipitators fly ash, which has a density in the range $2.0-2.5 gcm^{-3}$, can improve various properties of selected matrix materials, including hardness, strength, and wear resistance, and reduce density. Cenosphere fly ash, which consists of hollow fly ash particles, can be used for the synthesis of ultra-light composite materials due to its low density, which ranges from $0.4-0.7 gcm^{-3}$ compared to metal densities. Is in Matrices, which range in the range $1.6-2.0 gcm^{-3}$ [23]. Coal fly ash has many uses [24] including cement additives, as masonry blocks, as concrete mixtures, as a material in lightweight alloys, as a concrete aggregate, as a flux, as a concrete aggregate, in roadways / runway construction, structural fillings are included. Materials, as roofing granules, and in grouting. The largest application of fly ash is in the cement and concrete industry, however, creative new uses for fly ash are

being actively sought, like the use of fly ash for the manufacture of MMC.

Fly-Ash

Fly ash is one of the residues generated in coal combustion. It is an industrial by-product recovered from the flue gas of coal-fired power plants. Depending on the source and makeup of the coal to be burned, the components of the fly ash produced vary greatly, but all fly ash contains substantial amounts of silica (silicon dioxide, SiO_2) (both amorphous and crystalline) and lime (calcium oxide, CaO). In general, fly ash has SiO_2 , Al_2O_3 , Fe_2O_3 as major components and Mg, Ca, Na, K, etc. as minor components. Fly ash particles are mostly spherical in shape and range from $1100\mu m$ to $100\mu m$ with a specific surface area, typically between 250 and $600 m^2 / kg$. The specific gravity of fly ash varies in the range 0.6–2.8 g/cc . The physical properties of fly ash mainly depend on the burning and burning conditions of coal. Class F fly ash is generally produced from high-rank (high carbon content) coals such as anthracite and bituminous coals, while class C fly ash is produced from low-grade coals. Fly ash particles are classified into two types, precipitators and cenospheres. Typically, solid spherical particles of fly ash are called precipitator fly ash and hollow particles of fly ash with density less than $1.0 gcm^{-3}$ are called cenosphere fly ash. A common type of fly ash is usually composed of crystalline compounds such as quartz, mullite and hematite, glassy compounds such as silica glass and other oxides. Precipitators fly ash, which has a density in the range $2.0-2.5 gcm^{-3}$, can improve various properties of selected matrix materials, including hardness, strength, and wear resistance, and reduce density. Cenosphere fly ash, which consists of hollow fly ash particles, can be used for the synthesis of ultra-light composite materials due to its low density, which ranges from $0.4-0.7 gcm^{-3}$ compared to metal densities. Is in Matrices, which range in the range $1.6-2.0 gcm^{-3}$. Coal fly ash has many uses including cement additives, as masonry blocks, as concrete mixtures, as a material in lightweight alloys, as a concrete aggregate, as a flux, as a concrete aggregate, in roadways / runway construction, structural fillings are included. Materials, as roofing granules, and in grouting. The largest application of fly ash is in the cement and concrete

industry, however, creative new uses for fly ash are being actively sought, like the use of fly ash for the manufacture of MMC.

Depending on the chemical composition:

Two classes of fly ash are defined by ASTM C618: Class F fly ash and Class C fly ash. The main difference between these classes is the amount of calcium, silica, alumina and iron content in the ash. The chemical properties of fly ash are largely influenced by the chemical content of burned coal (i.e., anthracites, bituminous and lignite).

1. Class F fly-ash

Burning of hard, old anthracites and bituminous coal usually produces Class F fly ash. This fly ash is pozzolanic in nature and contains less than 13% lime (CaO). Considering the pozzolanic properties, the glassy silica and cementum of class F fly ash requires cement agents such as Portland cement, quick Crete, or hydrated lanas with the presence of water to inactivate and produce cemented compounds. Alternatively, addition of a chemical admixture such as sodium silicate (water glass) to Class F layers may lead to the formation of a geofilmer.

2. Class C fly-ash

In addition to pozzolanic properties, fly ash produced by the burning of young lignite or sub-bituminous carleae also has some self-cementing properties. In the presence of water, Class C fly ash will become stiff and strong over time. Class C fly ash typically contains more than 20% lime (CaO). Unlike Class F, self-cementing Class C fly ash does not require an activator. The content of alkali and sulfate (SO_4) is generally higher in Class C fly ash.

Based on size, shape, and structure:

1. Precipitator Fly-Ash

It is spherical in nature; the spheres are solid, and the density is in the range of $2.0-2.5 \text{ gcm}^{-3}$.

2. Cenosphere Fly-Ash

It is also spherical in shape, but they are spherical in color, so the density of such fly ash is much lower than

that of the precipitator fly ash. The density here is less than 1 gram cm^{-3} ($0.3-0.6 \text{ gcm}^{-3}$).

2.9 Why Fly-Ash?

1. The preference for using fly ash as a filler or consolidant in metal and tool matrices stems from the fact that fly ash is a by-product of coal combustion that is abundant (80 million tons per year) and inexpensive. Because it is excessive. Currently, the land is undeveloped. At the moment, only a limited amount of equipment for manufacturing glass microphases is in use, mostly due to the high cost of production. Thus, by adding fly ash into polymer and metallic alloy matrixes, the material cost of the composites has been dramatically decreased. However, little information is available to aid in the design of composite materials, with the exception of a single attempt to incorporate fly ash into adaptor and metal matrix composites. Cenosphere fly ash has a density somewhat more than that of talc and calcium carbonate, but slightly less than that of glass. Cenosphere is anticipated to be far less expensive than glass. In terms of cost per cell, the nanosphere may be one of the least expensive fillers.
2. Fly-ash has a high electrical resistivity, a low thermal conductivity, and a low density, which can be advantageous when fabricating light-weight platelet composites.
3. Using fly ash as a filler in aluminium castings lowers costs, increases density, and improves cough, wear, and abrasion resistance [23]. This results in an increase in machinability, shago capacity, coefficient of coefficient, and other properties that are required in a variety of industries, including automotive.
4. Because the production of aluminium is decreased by the utilization of fly ash. This minimizes greenhouse gas emissions associated with bauxite and blackminia extraction.

VII. Experiments details

The purpose of this study was to determine whether fly-ash could be used as a low-cost adsorbent for the removal of metal ions from mine wastewater. To optimize the adsorption capacity of coal fly-ash samples, various modification procedures were designed, and the results compared. We characterized samples of pristine and modified coal fly-ash and compared their adsorption properties, as well as their properties and preparation issues. Further research was conducted to determine the optimal preparation. To eliminate the possibility of metal hydroxide formation during adsorption studies, the fly-ash system was examined alone at pH conditions and in the presence of each metal ion solution. Other adsorption properties were investigated, including adsorption amount, adsorption kinetics, and the effect of temperature on adsorption. Additionally, the maximum adsorption

capacity of a given adsorbent for each metal ion was determined in both synthetic and mine water samples. Additionally, the adsorption kinetic model was investigated. Additionally, column studies and adsorption/desorption cycles of fly-ash were conducted to determine the material's potential application. Additionally, preliminary studies on fly-ash powder moulding were conducted to increase its applicability in real-world environmental systems.

To begin, 400 g of commercially pure Aluminum was melted and poured into an earthen graphite crucible in a resistant hot muffle furnace. This was accomplished by raising the melt temperature to 993K and purifying the hexachloroethane industries. Then, using a mixed casting process, fire-fly ash (13%) was prepared. We used 400 grammes of commercially pure active and 52 grammes of fly ash to accomplish this. To remove moisture from the fly-ash particles, they were preheated to 373K for two hours. The commercially pure user was melted at 993K and the hexachloroethane brands were purified. It was then melted with the assistance of a mild steel stirrer. The melt created as a result of the melting of fly-ash particles. While adding particles, the molten temperature was maintained between 953K and 993K. The crucible was then coated with melted clay graphite. Fly-ash was analyzed for particle size and chemical composition. Both commercially pure aluminium and an Al – 13 percent fly-ash composite were subjected to defect testing and density assessments. A Brinell defect testing machine with a 500 Kg load and a 10 mm diameter steel ball indent was used to determine the injured bone. The defect measurement detention duration was set to 30 seconds.

A wear testing machine was used to investigate the wear properties of a commercially pure aluminium and aluminum-13 percent fly-ash composite. Cylindrical samples with a diameter of 1 cm and a length of 2.1 cm were made from cast active and an Al – 13 percent fly-ash composite. For ten minutes, it was tested at 68.68 N load and 500 rpm. Both were analyzed using SEM and EDS.

Worked:

1. Al was melted and poured commercially pure.

2. Stir casting was used to make an Al – 13% fly-ash composite.
3. Fly ash was analysed chemically.
4. A particle size study was conducted to determine the suitability of fly ash for application.
5. Density and neural measurements were made on commercially pure aluminium samples as well as on Al – 13 percent fly-ash composite samples.
6. We examined and compared the wear parameters of commercially pure aluminium and Al – 13 percent fly-ash composites.
7. Both learnings were subjected to SEM analysis.
8. For both microanalyses of the material, EDS was used.

VIII. Conclusion

Alternative methods for making aluminium matrix composites have been identified, with the stir casting method receiving particular interest. This method briefly discusses the weight or volume fraction of Al Fly-Ash in the composite, the Al Fly-Ash particulate size, the Al Fly-Ash preheating temperature, the preheating and melting temperature of Al matrix, the stirring speed all through agitation of the composite liquid phase, the use of flux, wetting agent, the preheating temperature of the mould, and the trying to pour temperature of the composite slurry into the mould.

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