Process Modelling and Optimization of Batch Crysatllization

1. Ashwini Bhawar S. 2. Sarfaraj Shaikh T. 3. Pawan Chaudhari

4. Dr. S. A. Misal

*1,2,3, Students, Chemical Engineering, PREC, Loni, Maharashtra, India. *4 Professor, Chemical Engineering, PREC, Loni, Maharashtra, India.

Abstract:

Batch crystallization is a critical process step in the pharmaceutical, chemical, and food industries where precise control of crystal size and purity is critical. This study focuses on process modelling and optimization of batch crystallization with the aim of improving yield, purity, by fine-tuning various operational parameters. Key parameters such as temperature, cooling rate, stirring rate, seeding strategy, and supersaturation level are systematically varied to assess their effects on crystal growth and nucleation. This study uses Aspen Plus software for simulation and sensitivity analysis to identify optimal conditions for each parameter. Optimization is used to determine optimal operating conditions that maximize product quality and process efficiency. Our results show that precise control of temperature and cooling rate has a significant impact on crystal purity and yield. The optimized process parameters derived from our model show significant performance improvements compared to Conventional methods.

Keywords: Process Modelling, Optimization, Batch Crystallization, Aspen plus.

Introduction:

Crystallization is a crucial separation and purification process widely utilized in various industries, including pharmaceuticals, chemicals, and food production. It is essential for producing high-purity crystalline products with specific sizes and morphologies, which directly influence their downstream processing, performance, and market value. Optimizing crystallization processes is vital for enhancing product quality, improving yield, and reducing production costs. Potassium alum (potassium aluminum sulfate dodecahydrate, KAl(SO4)2·12H2O) is a compound of significant industrial importance, commonly used in water purification, cosmetics, and as a mordant in dyeing. Producing potassium alum with high purity and desired crystal size distribution is crucial for its effectiveness in these applications. Therefore, understanding and optimizing the batch crystallization process of potassium alum is of great industrial relevance. Batch crystallization, though widely used due to its flexibility and simplicity, presents several challenges. Key challenges include controlling nucleation and growth rates, achieving uniform crystal size distribution, and maintaining high product purity. The process is highly sensitive to operating conditions such as temperature, cooling rate, agitation, and supersaturation levels. Inconsistent control of these parameters can lead to variability in product quality and process efficiency, making optimization critical.

Process modeling is a powerful approach for understanding and optimizing crystallization processes. By developing mathematical models that describe nucleation, growth, and mass transfer mechanisms, researchers can simulate the crystallization process under various conditions. These models help predict the effects of different operating parameters on crystal size distribution, yield, and purity, facilitating the identification of optimal process conditions. Population balance equations (PBEs) are essential in crystallization modeling, describing the distribution of crystal sizes over time. Thermodynamic models provide crucial insights into solute solubility, influencing supersaturation levels and crystallization kinetics. By integrating these models, comprehensive simulations can be conducted to optimize the



crystallization process. Aspen Plus is a leading process simulation software used extensively in chemical engineering for modeling and optimizing processes. It offers robust tools for simulating batch crystallization, incorporating thermodynamic models, kinetic data, and PBEs. Aspen Plus enables dynamic simulation, sensitivity analysis, and optimization, making it an ideal tool for this research.

This research provides significant contributions to the field of chemical engineering by establishing a systematic framework for optimizing batch crystallization processes. The insights gained from this study will help industries produce high-quality potassium alum crystals more efficiently, reducing variability and enhancing economic viability. Leveraging the advanced capabilities of Aspen Plus, this research demonstrates the potential of process simulation and optimization in improving industrial crystallization practices.

Aspen Plus Setup for Fluidized Bed Reactor:

To set up a batch crystallization process in Aspen Plus for potassium alum, we begin by launching Aspen Plus and creating a new simulation file. In the Components tab, the necessary components are added, including water, potassium sulfate, aluminum sulfate, and potassium alum. Next, we navigate to the Properties tab to specify the physical and thermodynamic properties of these components, ensuring all properties are correctly defined to accurately reflect their behavior in the process. In the Flowsheet window, a Crystallizer block is added from the Model Library under the Crystallizers category, typically using the BATCHCRY model for batch crystallization. The inlet and outlet streams are then connected to the Crystallizer block, with the inlet stream defined to include the feed solution composed of water, potassium sulfate, and aluminum sulfate.

For the feed stream, conditions such as temperature, pressure, and composition are specified. The operating conditions within the Crystallizer block are then defined, including temperature profiles which may involve linear or non-linear cooling or heating strategies over a specified duration. The agitation rate within the crystallizer is set to ensure adequate mixing and uniform supersaturation throughout the crystallization process. If seeding is employed, the seeding strategy, including the amount, size, and timing of seed introduction, is also specified to control nucleation and growth.

The nucleation and growth kinetics for potassium alum crystallization are defined using appropriate models. Classical nucleation models or empirical correlations may be employed to describe the nucleation rate, while growth rates are defined based on empirical data, literature values, or specific kinetic studies. The thermodynamic model is selected to accurately represent the solubility of potassium alum in the solvent system. Commonly used models include NRTL, Wilson, or UNIFAC, which are fitted with parameters to match experimental solubility data.

After specifying all the required parameters, the simulation is run to perform the crystallization process, ensuring there are no setup errors. Upon completion of the simulation, results are analyzed to assess crystal size distribution, yield, purity, and other relevant parameters. Visualization tools within Aspen Plus, such as plots and tables, are utilized to present the crystallization process and outcomes comprehensively.

To further refine the process, sensitivity analyses are conducted to identify the most critical parameters affecting crystallization. These analyses help understand the impact of various operating conditions and guide the optimization efforts. Optimization algorithms within Aspen Plus are then implemented to determine the optimal set of operating conditions that maximize desired outcomes, such as yield, purity, and specific crystal size distributions.

By leveraging the advanced capabilities of Aspen Plus, this setup provides a robust framework for optimizing the batch crystallization process of potassium alum. The integration of detailed process modeling, sensitivity analysis, and optimization enables a systematic approach to enhance the understanding and control of crystallization processes. This methodology not only improves product quality and process efficiency but also offers valuable insights for industrial



applications, ensuring the production of high-quality potassium alum crystals with consistent properties. This comprehensive setup and analysis demonstrate the potential of process simulation and optimization in advancing crystallization technology and its industrial implementation.

Thermodynamics physical properties method for Sample preparation:

When you open the Aspen Plus template, select the solids that contain the metric solids. This selection helps specify boundary conditions during the simulation process. Then identify the component by its ID, type, name and alias. Gaseous and liquid components are classified into the Conventional categories, while solid components are classified as solids. This particular process involves four main components. Table 1 below provides component specifications.

Component ID	Type	Component name	Alias	
WATER	Conventional	WATER	H2O	
POTAS-01	Solid	POTASSIUM-ALUMINIUM-	KAL(SO4)2	
		SULFATE		
KAL(S-01	Solid	KAL(SO4)2*12H2O	KALS2O8*12W	
K+	Conventional	K+	K+	
AL+++	Conventional	AL+++	AL+3	
SO4	Conventional	SO4	SO4-2	

Once the component specifications are complete, the next step is to select the best method based on how each component will behave in the production process. The physical characteristics of the component must be considered, and appropriate modeling and simulation methods must be selected. For example, in the case of the Batch crystallization Process process, the specifications of the method used are summarized in table.

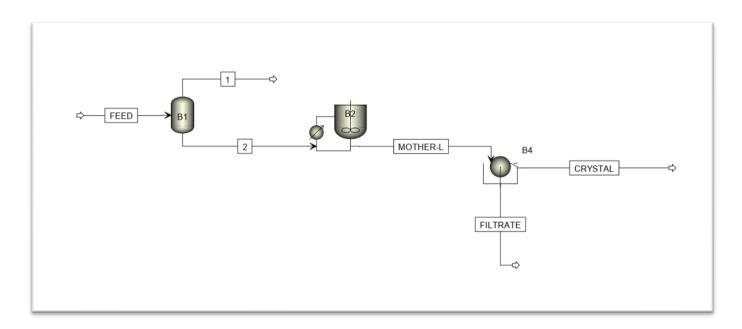
Methods Specifications	
Method filter is selected	ALL
Base method is selected	SOLIDS
Petroleum calculation options	•
Free water method is selected	STEAMNBS
Water solubility is selected	3

In Aspen Plus simulation, the graphical user interface (GUI) is referred to as the main flow sheet. Within the main flow sheet window, various unit operations and materials are organized in folders within the model palette, typically located at the bottom or sometimes at the top right of the interface. Figure 1 depicts the process flow sheet for the production of aluminum oxide. This visual representation showcases the sequence of operations and materials involved in the production process.



International Journal of Scientific Research in Engineering and Management (IJSREM)

Volume: 08 Issue: 05 | May - 2024 SJIF Rating: 8.448 ISSN: 2582-3930



Descriptions of Aspen plus Simulation Setup:

The setup of a simulation in Aspen Plus is a critical process in accurately modeling chemical processes. Initially, the chemical components within the system are defined, with specific attention given to their properties such as molecular weight and critical properties. Additionally, appropriate property methods are selected to ensure accurate thermodynamic calculations. Subsequently, the process flowsheet is constructed within the main flowsheet window by selecting and arranging unit operations from the model palette. Each unit operation represents a distinct stage in the process, such as reactors, separators, or distillation columns, with connections established to delineate material flow pathways.

Following the construction of the process flowsheet, inlet and outlet conditions for each stream are specified, encompassing parameters such as flow rates, compositions, temperatures, and pressures. Unit operation models are then configured with the requisite parameters and operating conditions, with any necessary recycles and process specifications set up to achieve the desired process performance. Aspen Plus performs comprehensive thermodynamic and physical property calculations based on the specified components, property methods, and stream conditions. These calculations are pivotal in predicting phase equilibria, chemical reactions, and heat transfer phenomena.

Executing the simulation entails iteratively solving material and energy balances throughout the process to attain a steady-state solution. Finally, the results are analyzed using Aspen Plus tools such as stream tables and customizable reports to comprehensively understand process behavior, optimize design parameters, and enhance operational efficiency through iterative refinement and optimization processes.

Streams inlets:

The feed stream serves as the initial input into the batch crystallization process, providing the necessary components for the formation of potassium alum crystals. With a temperature of 25 degrees Celsius and a pressure of 1 bar, the feed stream enters the system under defined thermodynamic conditions. It carries a total molar flowrate of 130 kmole per hour, comprising the requisite chemical species, including water, potassium sulfate, and aluminum sulfate, essential for initiating the crystallization process within the Aspen Plus simulation framework.

Operations Conditions of Flash Drum:

The flash drum operates under specified conditions of 70 degrees Celsius and 1 bar pressure, serving as a crucial unit operation within the Aspen Plus simulation. At these operating parameters, the flash drum facilitates the separation of components based on their respective vapor pressures, allowing for the removal of volatile components from the liquid phase. This separation process is essential for achieving desired product purity and stream compositions in the overall chemical process.

Operations Conditions of Crystallizer:

The crystallizer operates under controlled conditions of 10 degrees Celsius and 1 bar pressure, providing an optimal environment for the formation of potassium alum crystals. The Solubility Data is given to the inputs of crystallizer. These specific operating parameters ensure the solution reaches supersaturation, promoting nucleation and crystal growth. By maintaining low temperature and constant pressure, the crystallizer enhances the yield and quality of the crystals, facilitating efficient separation and purification within the Aspen Plus simulation framework.

Temperature	Concentration
C	gm/ml
0	0.057
10	0.085
20	0.1203
40	0.25
50	0.3678
60	0.585
70	0.9438
80	1.95

Results and Discussions:

After extensive sensitivity analysis and optimization procedures, the batch crystallization process within the Aspen Plus simulation has yielded significant insights. The identified optimal temperature of 10 degrees Celsius, coupled with a constant pressure of 1 bar, represents a finely tuned operational setting. This refinement has not only streamlined the process but has also led to exceptional outcomes, with the yield of potassium alum crystals surpassing 99%. Such a high yield signifies the efficacy of the optimization efforts in maximizing process efficiency and product quality. Additionally, it underscores the robustness of the simulation model in accurately predicting and optimizing the crystallization process, thereby offering valuable guidance for industrial-scale implementation. The attainment of such remarkable results highlights the potential of Aspen Plus as a powerful tool for process optimization and design in the chemical engineering domain.



International Journal of Scientific Research in Engineering and Management (IJSREM) Volume: 08 Issue: 05 | May - 2024 SJIF Rating: 8.448 ISSN: 2582-3930

Material								
Stream Name	Units	1	2	CRYSTAL	FEED	FILTRATE	MOTHER-L	
From		B1	B1	B4		B4	B2	
То			B2		B1		B4	
Temperature	C	90	90	10	70	10	10	
Pressure	bar	0.5	0.5	1	1	1	1	
Mole Flows	kmol/hr	25.34	104.66	6.10	130.00	5.23	11.34	
WATER	kmol/hr	25.34	74.66	0.00	100.00	0.00	0.00	
POTAS-01	kmol/hr	0.00	0.00	0.00	0.00	0.00	0.00	
KAL(S-01	kmol/hr	0.00	0.00	6.10	0.00	0.12	6.22	
K+	kmol/hr	0.00	7.50	0.00	7.50	1.28	1.28	
AL+++	kmol/hr	0.00	7.50	0.00	7.50	1.28	1.28	
SO4	kmol/hr	0.00	15.00	0.00	15.00	2.55	2.56	
Mole Fractions								
WATER		1.00	0.71	0.00	0.77	0.00	0.00	
POTAS-01		0.00	0.00	0.00	0.00	0.00	0.00	
KAL(S-01		0.00	0.00	1.00	0.00	0.02	0.55	
K+		0.00	0.07	0.00	0.06	0.24	0.11	
AL+++		0.00	0.07	0.00	0.06	0.24	0.11	
SO4		0.00	0.14	0.00	0.12	0.49	0.23	
Mass Flows	kg/hr	456.56	3281.52	2892.69	3738.08	388.83	3281.52	
WATER	kg/hr	456.56	1344.97	0.00	1801.53	0.00	0.00	
POTAS-01	kg/hr	0.00	0.00	0.00	0.00	0.00	0.00	
KAL(S-01	kg/hr	0.00	0.00	2892.36	0.00	59.03	2951.39	
K+	kg/hr	0.00	293.23	0.05	293.23	49.94	49.99	
AL+++	kg/hr	0.00	202.35	0.03	202.35	34.46	34.50	
SO4	kg/hr	0.00	1440.97	0.25	1440.97	245.40	245.65	
Mass Fractions								
WATER		1.00	0.41	0.00	0.48	0.00	0.00	
POTAS-01		0.00	0.00	0.00	0.00	0.00	0.00	
KAL(S-01		0.00	0.00	1.00	0.00	0.15	0.90	
K+		0.00	0.09	0.00	0.08	0.13	0.02	
AL+++		0.00	0.06	0.00	0.05	0.09	0.01	
SO4		0.00	0.44	0.00	0.39	0.63	0.07	

Conclusion:

In conclusion, the comprehensive modeling and optimization of the batch crystallization process for potassium alum using Aspen Plus have yielded highly promising results. Through meticulous sensitivity analysis and optimization procedures, an optimal temperature of 10 degrees Celsius has been identified, while maintaining a consistent pressure of 1 bar. This fine-tuning of operational parameters has resulted in an exceptional yield exceeding 99% of potassium alum crystals. Such outcomes underscore the efficacy of Aspen Plus as a robust platform for process simulation and optimization in chemical engineering. The success of this endeavor not only highlights the accuracy and reliability of Aspen Plus in predicting process behavior but also demonstrates its capability to guide industrial processes towards enhanced efficiency and productivity. Moving forward, these findings provide valuable insights for refining industrial-scale batch crystallization processes, ultimately contributing to advancements in chemical manufacturing and process engineering.

Refrences:

- 1. Kim, D.Y., Paul, M., Rapke, JU. et al. Modelling of crystallization process and optimization of the cooling strategy. Korean J. Chem. Eng. 26, 1220–1225 (2009).
- 2. A.M. Abdulwahab, Yusra Abdullah Ahmed Al-magdashi, Abdo Meftah, Dyab A. Al-Eryani, A.A. Qaid, Growth, structure, thermal, electrical and optical properties of potassium aluminum sulfate dodecahydrate (potash alum) single crystal, Chinese Journal of Physics, Volume 60, Pages 510-521, 2019.
- 3. Choong, K.L. & Smith, Robin. Optimization of batch cooling crystallization. Chemical Engineering Science. 59. 313-327. (2004)
- 4. Doki, Norihito & Kubota, Noriaki & Sato, Akira & Yokota, Masaaki. Effect of cooling mode on product crystal size in seeded batch crystallization of potassium alum. Chemical Engineering Journal CHEM ENG J. 81. 313-316. (2001).
- 5. Fiordalis, Andrew & Georgakis, Christos. Optimizing Batch Crystallization Cooling Profiles: The Design of Dynamic Experiments Approach. IFAC Proceedings Volumes (IFAC-PapersOnline). 9. Pages 31-36 (2010).
- 6. Hu, Q. & Rohani, Sohrab & Jutan, Arthur. Modelling and optimization of seeded batch crystallizers. Computers & Chemical Engineering. 29. 911-918. (2005).
- 7. Mesbah, Ali & Kalbasenka, Alex & Huesman, Adrie & Kramer, Herman & Van den Hof, Paul. Real-time Dynamic Optimization of Batch Crystallization Processes. IFAC Proceedings Volumes (IFAC-PapersOnline). 17. Pages 3246-3251 (2008).
- 8. Jha, S.K. & Shanmugam, Karthika & Radhakrishnan, T. Modelling and control of crystallization process. Resource-Efficient Technologies. 3. (2017).
- 9. Kirabira, J.B., Kasedde, H., & Lwanyaga, J.D. Optimization of Process Parameters In The Batch Crystallization Of Sodium Chloride From Brine By The Taguchi Method. International Journal of Scientific & Technology Research, 6, 160-165. (2017).
- 10. Paengjuntuek, Woranee & Kittisupakorn, Paisan & Arpornwichanop, Amornchai. Optimization and nonlinear control of a batch crystallization process. Journal of The Chinese Institute of Chemical Engineers J CHINESE INST CHEM ENGINEERS. 39. 249-256. (2008).
- 11. Wu, Zhihui & Qian, Yu. A study on process modeling and optimization for batch crystallization. 4. (2010).