

# The Role of Non-ionic Surfactants in Modern Textile Manufacturing: A Comprehensive Review

Harshal Patil; Jyotsna Waghmare

Department of Oils, Oleochemicals and Surfactant Technology Institute of Chemical Technology (ICT), Nathalal Parikh Marg, Matunga (E), Mumbai-400019, Maharashtra. India.

## ABSTRACT: -

Non-ionic surfactants have emerged as crucial agents in textile production because of their unique physicochemical characteristics, such as neutrality, low toxicity, and good compatibility with various chemicals and fibres. Their amphiphilic nature allows them to wet, emulsify, disperse, and stabilize well, making them suitable for application at different stages of textile processing especially in pre-treatment, dyeing, and finishing. This review discusses the structural diversity of non-ionic surfactants and examines their functionality in enhancing fabric quality, dye uniformity, process efficiency, and environmental compliance. Individual surfactant types like alcohol ethoxylates, fatty acid ethoxylates, sugar-based surfactants, and block copolymers are explored for their individual contributions to textile treatment. In addition, the paper discusses new developments such as bio-based surfactants, thermos responsive materials, and nanostructured finishes that broaden the application of non-ionic surfactants in smart textiles and green manufacturing. Environmental factors such as biodegradability, aquatic toxicity, and regulatory limitations are also discussed. By summarizing available literature and emphasizing new trends, the review of this article should present an integral view of the mechanism, advantage, and future outlook of non-ionic surfactants in contemporary textile production. The extended progress of non-ionic surfactants will be crucial to address the requirements of the industry for high performance, sustainability, and process innovation.

**KEYWORDS:** - Non-ionic surfactants, textile processing, dyeing auxiliaries, eco-friendly surfactants, textile pre-treatment, finishing agents,

## 1. INTRODUCTION

Non-ionic surfactants have become integral materials in numerous industries, most notably in textile production, based on their singular attributes and flexible applications. Such chemicals, known for their non-charged status, are an integral part of various processes in the textile manufacturing process. From production of fibres to finishing of the fabric, non-ionic surfactants provide excellent wetting, emulsifying, and dispersing properties, and thus find themselves being crucial in dyeing, printing, and treatment of fabric. Non-ionic surfactants feature no charged groups within their molecular structure.[1], [2] Instead, they have hydrophilic (water-attracting) and hydrophobic (water-repelling) parts, and thus can act between water and oil. This biophilic character renders them very effective in numerous textile processes, such as scouring, bleaching, dyeing, and finishing. In addition, Non-ionic surfactants are less toxic and more biodegradable than their anionic and cationic counterparts, thus a choice of preference in environmentally friendly textile processing. These surfactants play a central role in processes like detergency, fabric treatment, and hydrophilic finishing. For example, Dehypon® LS54, an ionic surfactant, is applied in wet cleaning in textile conservation, though its low cloud point could present cleaning difficulties [3]. Moreover, new non-ionic surfactants, for example, those with cyclic structure and hydroxyl groups, have been synthesized for fabric treatment with improved properties such as increased solubility and stability [4]. With respect to environmental influence, non-ionic surfactants such as alkyl Nonylphenol ethoxylated (APEO) are tracked in textile wastewater owing to their possible ecological impact, though they are less toxic than anionic surfactants [5]. Additionally, non-ionic surfactants are applied in surfactant formulations for the provision of a permanent hydrophilic finish to textiles, frequently in conjunction with anionic or cationic surfactants and hydrophobic additives such as fatty acid esters. These uses focus on the utility of non-ionic surfactants in texturing properties in addition to

controlling environmental issues and thus are essential in the push by the textile industry to counterbalance performance versus sustainability. They are widely used in the synthesis of polymeric nanoparticles, emulsions, and stabilizing carriers in liquid media. They possess special properties like low toxicity, good stability, and compatibility with other chemicals, which make them unavoidable in the textile sector. Non-ionic surfactants are more effective in the wetting, dyeing, and finishing of textiles, resulting in better fabric quality and performance. This detailed review seeks to investigate the multifaceted applications, mechanisms, and recent developments of non-ionic surfactants in textile production, noting their major influence on the industry's efficiency and sustainability [1]. They are most favored in industrial applications because of their non-toxicity, formulation ease, and cost effectiveness [6]. Non-ionic surfactants help in the formulation of eco-friendly processes for textile production to fulfil consumer demands. The NMMO-based lyocell process using non-ionic surfactants is of great promise in meeting environmental and customer demands in producing fibres [7]. Besides, non-ionic surfactants having ethylene oxide (EO) head groups also show thermal solubility and can be used to develop temperature-insensitive formulations when blended with sucrose and glucose-based surfactants. Non-ionic surfactants are extremely essential in improving the sustainability of textile production processes and consumer demands. Their utility goes beyond the NMMO-based lyocell process, since they may be applied in any step of textile manufacturing, such as preparation of fibres, dyeing, and finishing. The thermoresponsive solubility of non-ionic surfactants containing ethylene oxide head groups also presents process control and efficiency benefits. [8] In addition, the complexation between polymers and non-ionic surfactants such as TX-100 has proved promising in enhancing hydrophilicity of textile substrates [9]. Non-ionic surfactants provide a number of benefits in textile production, such as improved dyeing efficiency and fabric quality. They have good wetting and foaming properties, which allow for better dye penetration and uniformity, especially with natural dyes, as evidenced by their synergism with different types of dyes [10]. Non-ionic surfactants are also less toxic compared to anionic surfactants, thus cutting environmental costs, an important aspect considering the pollution potential of the textile industry [5]. However, their disadvantages are their lower cloud points, which may make wet cleaning processes difficult, as with Dehypon® LS54, and reduce their effectiveness under some conditions [5]. Additionally, although they enhance color fastness, overall performance may be concentration and pH level dependent during dyeing [11]. Therefore, although non-ionic surfactants are advantageous, proper evaluation of their properties and application conditions is necessary. [12] With the fast-paced development of the textile manufacturing industry, non-ionic surfactants have become indispensable chemical entities that play a major role in multiple steps of textile processing. These multifaceted compounds, defined by their neutral charge and distinct molecular structure, have transformed conventional textile manufacturing processes by optimizing process efficiency, product quality, and environmental compatibility. The increased demand for high-performance textiles and stricter environmental legislation have fueled the interest in researching how to better understand and maximize the use of non-ionic surfactants in textile processing. The textile sector is one of the largest and most diversified sectors in the world, encompassing a lot of chemical processing involving the application of surfactants. [13-14] Surfactants, also known as surface-active agents, are chemicals that reduce the surface tension between two liquids or between a liquid and a solid. Surfactants find extensive application within the textile industry due to their capacity to enhance wetting, emulsification, dispersion, and detergency. Amongst the different categories of surfactants, Non-ionic surfactants have been of great interest because of their special characteristics, including low foaming, hard water stability, and compatibility with other chemicals. This review paper discusses the basic principles, present applications, and future trends in the use of non-ionic surfactants in the textile industry. We investigate their significant function in important processes like scouring, dyeing, finishing, and fabric care, and study their influence on fabric properties, process parameters, and environmental aspects. This in-depth review reviews the basic principles, existing applications, and future trends in the use of non-ionic surfactants in the textile manufacturing industry. We discuss their vital function in important processes like scouring, dyeing, finishing, and fabric care, examining their influence on fabric properties, process parameters, and environmental factors. By emphasizing both their chemical flexibility and ecological advantages, this review highlights the promise of non-ionic surfactants to address the increasing needs for efficiency, innovation, and sustainability in textile processing.

## 2. Chemical Structure and Properties of Non-ionic Surfactants

Non-ionic surfactants are amphiphilic molecules defined by their absence of electrical charge in the hydrophilic section of the molecule. Their molecular structure is usually a hydrocarbon chain (hydrophobic tail) and a non-charged hydrophilic head group, typically made up of chains of polyoxyethylene (POE) or polyhydroxy compounds. This distinctive molecular

design allows non-ionic surfactants to lower the surface tension at interfaces and allow dissimilar phases, e.g., oil and water, to interact [15]. The lack of ionic charge makes these surfactants less electrolyte and water hardness sensitive than their ionic counterparts and yet remain operational over a broad range of pH values. The majority of non-ionic surfactants show an inverse temperature and solubility relationship, being less soluble at higher temperatures as their ethoxylate chains dehydrate. This phenomenon, referred to as the cloud point, is an important formulation design parameter for textile processing. Non-ionic surfactants are usually noted to exhibit excellent compatibility with other surfactant classes, allowing for synergistic blends that maximize performance in a wide range of textile uses [16]. The most important difference between non-ionic and ionic surfactants is in their hydrophilic head group constitution and subsequent charge properties. Anionic surfactants have negatively charged head groups, usually sulphates, sulfonates, or carboxylates, which make them very effective against particulate soil and oils but sensitive to water hardness and acid conditions. Their intense electrostatic repulsions give them very good detergents but possible severity against protein fibres such as wool and silk. Cationic surfactants, having positively charged quaternary ammonium groups, are poor detergents but good softeners and antistatic because they adhere to the negatively charged surface of textiles. They are incompatible with anionic surfactants by virtue of their positive charge and restrict the formulator to some choices. Amphoteric surfactants have both an anionic and a cationic charge, with performance varying according to solution pH; they provide very good mildness and compatibility with other classes of surfactant.[17] Non-ionic surfactants, on the other hand, exhibit stable performance over ranges of changing pH and water hardness values, show good compatibility with all classes of surfactant, and tend to exhibit better mildness on sensitive fibres. They perform well under low-temperature conditions where their solubility is optimized. Though generally more costly than their anionic counterparts, non-ionic surfactants typically exhibit better overall performance in difficult-to-handle textile processing situations, especially where electrolyte tolerance, hard water stability, and soft handling of fibres are critical concerns.[18]

Sr.no	Surfactant Type	Chemical Structure	Environmental Profile
1	<b>Alcohol Ethoxylates</b>	$R-(OCH_2CH_2)_n-OH$ (R = C8-C18, n = 3-20)	<ul style="list-style-type: none"> <li>• Good biodegradability</li> <li>• Linear types more eco-friendly</li> <li>• Moderate aquatic toxicity</li> </ul>
2	<b>Alkylphenol Ethoxylates</b>	$C_9H_{19}-C_6H_4-(OCH_2CH_2)_n-OH$ (for NPEs)	<ul style="list-style-type: none"> <li>• Poor biodegradability</li> <li>• Restricted/banned in many regions</li> <li>• Estrogenic degradation products</li> </ul>
3	<b>Fatty Acid Ethoxylates</b>	$R-COO-(CH_2CH_2O)_n-H$	<ul style="list-style-type: none"> <li>• Moderate to good biodegradability</li> <li>• Lower aquatic toxicity</li> </ul>
4	<b>Fatty Alcohol Ethoxylates</b>	$R-(OCH_2CH_2)_n-OH$ (from natural sources)	<ul style="list-style-type: none"> <li>• Good biodegradability</li> <li>• Derived from renewable feedstocks</li> <li>• Moderate aquatic impact</li> </ul>
5	<b>Block Copolymers</b>	$HO-(EO)_x-(PO)_y-(EO)_x-H$	<ul style="list-style-type: none"> <li>• Slow but complete biodegradation</li> <li>• Low toxicity</li> <li>• Persistent in some environments</li> </ul>
6	<b>Sugar-based Surfactants</b>	$R-O-(glucose)_n$ (for APGs)	<ul style="list-style-type: none"> <li>• Excellent biodegradability</li> <li>• Low aquatic toxicity</li> <li>• Sustainable sourcing possible</li> </ul>

7	<b>Silicone Surfactants</b>	Polysiloxane with EO/PO side chains	<ul style="list-style-type: none"> <li>Limited biodegradability</li> <li>Persistent in environment</li> <li>Low aquatic toxicity</li> <li>Higher production footprint</li> </ul>
8	<b>Biosurfactants</b>	Various (glycolipids, lipopeptides, phospholipids, etc.)	<ul style="list-style-type: none"> <li>Excellent biodegradability</li> <li>Minimal environmental impact</li> <li>Carbon-neutral potential</li> <li>Non-toxic to aquatic organisms</li> </ul>

Table 1 Non-ionic Surfactant & Their types



Figure 1 Overview of Non -ionic surfactant types

Sr.no	Surfactant Type	Key Properties	Primary Textile Applications
1	<b>Alcohol Ethoxylates</b>	<ul style="list-style-type: none"> <li>Excellent wetting</li> <li>Low foaming at high temps</li> <li>Versatile HLB range</li> <li>Good detergency</li> </ul>	<ul style="list-style-type: none"> <li>Scouring processes</li> <li>Softening formulations</li> <li>Dyeing wetting agents</li> <li>Hydrophobic dispersants</li> </ul>
2	<b>Alkylphenol Ethoxylates</b>	<ul style="list-style-type: none"> <li>Excellent alkaline stability</li> <li>Cost-effective</li> <li>High performance</li> <li>Effective emulsifiers</li> </ul>	<ul style="list-style-type: none"> <li>Wool scouring</li> <li>Emulsifiers</li> <li>Dye/pigment dispersants</li> <li>Stain removing</li> </ul>
3	<b>Fatty Acid Ethoxylates</b>	<ul style="list-style-type: none"> <li>Superior emulsification</li> <li>Acid stability</li> <li>Low foam</li> <li>Effective at low concentrations</li> </ul>	<ul style="list-style-type: none"> <li>Fabric softeners</li> <li>Crease-resistant finishing</li> <li>Fiber lubricants</li> <li>Dyeing auxiliaries</li> </ul>
4	<b>Fatty Alcohol Ethoxylates</b>	<ul style="list-style-type: none"> <li>Excellent detergency</li> <li>Good wetting</li> <li>Wide temperature efficacy</li> <li>Customizable properties</li> </ul>	<ul style="list-style-type: none"> <li>Detergent formulations</li> <li>Dyeing levelling agents</li> <li>Water repellent finishes</li> <li>Printing dispersants</li> </ul>

5	<b>Block Copolymers</b>	<ul style="list-style-type: none"> <li>Controlled foam</li> <li>Temperature-responsive</li> <li>Good compatibility</li> <li>Excellent stabilization</li> </ul>	<ul style="list-style-type: none"> <li>Antifoaming in jet dyeing</li> <li>Difficult dye dispersants</li> <li>High-speed process regulators</li> <li>Functional finish emulsifiers</li> </ul>
6	<b>Sugar-based Surfactants</b>	<ul style="list-style-type: none"> <li>Excellent biodegradability</li> <li>Renewable sourcing</li> <li>Mild to skin and fibers</li> <li>Compatible with other surfactants</li> </ul>	<ul style="list-style-type: none"> <li>Eco-friendly detergents</li> <li>Delicate fiber processing</li> <li>Bio-scouring</li> <li>Dermatologically friendly treatments</li> </ul>
7	<b>Silicone Surfactants</b>	<ul style="list-style-type: none"> <li>Extreme surface tension reduction</li> <li>High thermal stability</li> <li>Superior spreading</li> <li>Effective at ultra-low concentrations</li> </ul>	<ul style="list-style-type: none"> <li>Water-repellent fabric wetting</li> <li>Foam control</li> <li>Premium softeners</li> <li>High-performance finishes</li> <li>Fiber lubricants</li> </ul>
8	<b>Biosurfactants</b>	<ul style="list-style-type: none"> <li>High biodegradability</li> <li>Low toxicity</li> <li>Functionality at extreme conditions</li> <li>Specificity in action</li> <li>Renewable sourcing</li> </ul>	<ul style="list-style-type: none"> <li>Eco-friendly textile processing</li> <li>Enzymatic desizing/scouring</li> <li>Natural antimicrobial finishes</li> <li>Sustainable dyeing processes</li> <li>Green detergency</li> </ul>

Table 2 Type of surfactant & their used in textile application

### 3. APPLICATION IN TEXTILE PROCESSING

#### 3.1 Applications in Textile Pre-treatment

Non-ionic surfactants are very important during textile pre-treatment operations because they possess excellent wetting, emulsifying, dispersing, and detergent characteristics. Unlike ionic surfactants, non-ionic surfactants bear no charge and thus exhibit high compatibility with different chemicals and low sensitivity to water hardness or pH levels. Typical examples include ethoxylated fatty alcohol, alkyl phenol ethoxylate and fatty acid ethoxylate. Such surfactants find perfect applications in textile processing where they are utilized during desizing, scouring, bleaching, and other preparatory steps to guarantee homogeneous and effective treatment of textiles. One of the major uses of non-ionic surfactants is in desizing, in which they assist in stripping sizing agents like starch, polyvinyl alcohol (PVA), and carboxymethyl cellulose (CMC) from textiles. Surfactants such as polyethylene glycol (PEG) esters and ethoxylated sorbitan esters (e.g., Tween series) enhance penetration of desizing enzymes or oxidizing agents by lowering surface tension so that efficient breakdown and stripping away of size material can be achieved. Their low-foaming character is especially useful in continuous processing, where foam can cause operation to be interrupted. [18-20] In scouring, non-ionic surfactants like fatty alcohol ethoxylates (e.g., laureth-7, steareth-10) and ethoxylated triglycerides are necessary for stripping natural impurities such as oils, waxes, and fats from cotton and other natural fibres. They serve as emulsifiers, disintegrating hydrophobic materials and dispersing them in the scouring bath for simple rinsing. Their capacity to perform well under alkaline as well as neutral conditions makes them suitable for various scouring recipes. In addition, non-ionic surfactants increase the wettability of textiles so that they can absorb subsequent chemicals evenly in dyeing and finishing treatments. [21-24]

Non-ionic surfactants also play an important role in bleaching operations, especially in hydrogen peroxide bleaching. Surfactants such as alkyl polyglucoside (APGs) and ethoxylated amines stabilize the bleaching bath by inhibiting decomposition of peroxide and ensuring uniform distribution of the bleaching agent on the fabric. Their mutual compatibility with oxidizing agents and enzymes makes them ideal for use in combined scouring-bleaching operations, allowing water and energy to be saved. Another significant use is in silk degumming, where non-ionic surfactants like ethoxylated castor oil (e.g., PEG-40 hydrogenated castor oil) help to remove sericin without harming the fragile silk fibres.

[21-23] They aid in emulsifying the gum residues, making them easier to remove while preserving the lustre and strength of the silk. Likewise, in scouring wool, surfactants such as block copolymers (e.g., poloxamers) and ethoxylated fatty acids help to remove grease and suint (natural wool grease) without fibre damage. Non-ionic surfactants are also employed in synthetic fibre processing, especially in the removal of spin finishes and lubricants used during yarn production. Some examples are ethoxylated fatty acids and alcohol alkoxyates, used to guarantee removal of the processing aids in their entirety, essential for ensuring uniform dye absorption. Additionally, non-ionic surfactants are environmentally friendly as opposed to certain ionic analogs because they tend to be biodegradable and less toxic. Some surfactants such as alkyl polyglucoside (APGs) and sugar-based ethoxylates are increasing in usage since they are environmentally friendly. Their resistance to high temperatures and hardness of water make them convenient for use in different textile pre-treatment processes.[19], [24-27]

### 3.2 Role of Non-Ionic Surfactants in Dyeing Processes

Non-ionic surfactants are crucial in textile dyeing as they enhance dye dispersion, facilitate uniform application of color, and increase process efficiency. They are chemically stable and compatible with various classes of dyes, making them a necessity in contemporary dyeing processes. Dye particle dispersion is one of their main roles, and surfactants like ethoxylated fatty alcohols (e.g., oleyl alcohol ethoxylate) and alkylphenol ethoxylates serve to disperse and break down dye aggregates for uniform distribution in the dye bath. This is specifically important in case of disperse dyes used in dyeing polyester, surfactants such as sorbitan monooleate (Span 80) and polysorbates (Tween series), ethoxylated fatty alcohol, ethoxylated triglyceride hinder the agglomeration of dye, decreasing chances of improper dyeing or spots. Dispersion mechanism involves attaching hydrophobic parts of surfactants onto dye particles as their hydrophilic ends dip into the water phase to stabilize the dispersion. This steric hindrance avoids re-aggregation of dye molecules during dyeing. For instance, nonylphenol ethoxylates with 8-10 units of ethylene oxide are especially good for disperse dyes, forming micelles that enclose dye particles and keep them in suspension even at high temperatures of 130-140°C employed in high-pressure dyeing of polyester.[28-29]

Non-ionic surfactants serve as levelling agents in achieving balanced dyeing by regulating the rate of absorption of dye. Some examples of these are block copolymers and fatty amine ethoxylates, which decelerate the migration of dyes, particularly in fibre-reactive and acid dyeing procedures. This helps to reduce blotchy dyeing and ensure greater color consistency. Levelling is accomplished by two major mechanisms: controlled retardation, where surfactants briefly complex with molecules of the dye to decrease their initial attraction to the fibre and enable more even distribution prior to fixation; and migration enhancement, where surfactants enable the transfer of molecules of the dye from zones of high concentration to zones of low concentration of the substrate. Advanced levelling agents such as EO/PO block copolymers show thermosensitive behaviour, with more enhanced retardation at lower temperatures and dye molecule release with increased temperature during the dyeing process. This temperature-responsive property is useful in the management of dye uptake kinetics in polyester dyeing and similar processes.[30-34] Non-ionic surfactants' anti-foaming action resolves problems in high-speed dyeing processes such as jet dyeing, which tend to create foam and promote uneven dyeing and machine overflow. Anti-foaming substances such as silicone-based polyether's and EO/PO (ethylene oxide/propylene oxide) copolymers are powerful defoamers that ensure stability in processes without influencing dyeing performance. The antifoam mechanism involves the destabilization of the air-water interface within foam bubbles, surface tension reduction at critical locations within the dyeing equipment, and the formation of hydrophobic films that disrupt foam structures. Modern antifoaming agents are designed specifically to be effective at the high temperatures (90-140°C) found in pressure dye systems as well as being compatible with the entire spectrum of auxiliary chemicals in the dye bath. [35-38]

One of the advantages of non-ionic surfactants is that they are compatible with a range of dye classes because they do not interact ionically with dyes. Ethoxylated castor oil enhances dye penetration and solubility for reactive dyes applied to cotton and rayon. Their multi-hydroxyl functionality offers very good compatibility with cellulosic substrates with an increase in dye diffusion into capillaries of fibres. They also serve to stabilize reactive groups under conditions of alkaline fixation. With wool and silk acid dyes, fatty alcohol ethoxylates help to distribute the dye evenly. Their mildness maintains the sensitive protein structure of these fibres while allowing level dyeing, with ethoxylates containing 12-15 carbon chains

and 7-9 EO units providing the best balance between dispersing power and fibre affinity. With disperse dyes applied to polyester, aryl ethoxylates improve dye dispersion at elevated temperatures. Their heat-stable architecture inhibits degradation under high-temperature dyeing, ensuring dispersion efficiency throughout the process and helping to facilitate thermomigration of dye molecules in the polymer architecture upon heat-setting. While dyeing with vat and sulfur dyes, PEG surfactants help reduce and solubilize dyes, assist in preserving water solubility of leucon forms, and inhibit premature reoxidation during the dyeing process.[35-39]

Non-ionic surfactants have a large impact on penetration and wetting by lowering surface tension of the dye baths from around 72 mN/m to 30-35 mN/m, which allows complete and fast wetting of substrates of textile. This aspect is particularly useful in the case of dense substrates such as denim or heavy twills with problems of penetration of dyes, natural fibres that have built-in hydrophobicity by waxes or contaminants, and synthetic fibre clumps where capillary forces inhibit access of dye. Ethoxylated alcohols of 10-13 HLB value have excellent wetting character, and on especially difficult substrates, advanced penetrating agents featuring a combination of non-ionic and amphoteric structures are used to cope with extreme hydrophobicity. Apart from these particular applications, non-ionic surfactants contribute to general dyebath stability by inhibiting precipitation of dyes or auxiliaries during hard water conditions, ensuring consistent dye solubility over temperature fluctuations, pH buffering against conditions that might degrade dye performance, and minimizing mechanical fibre-to-fibre friction during processing. The advantages are reflected in increased process efficiency, decreased reject rates, and lower water and energy consumption in advanced dyeing processes. Further, newer generations of non-ionic surfactants are formulated with better biodegradability profiles to cater to environmental issues related to textile processing.[40-42]

### 3.3 Finishing Applications

Non-ionic surfactants are also important as emulsifiers in the application of hydrophobic finishing agents like silicones, waxes, and water repellents. They disperse these substances in water so that they can be applied evenly to the fabric. Their stability over a broad pH range and electrolyte resistance makes them ideal for many textile formulations. In silicone finishes, for instance, non-ionic surfactants facilitate the cleavage of silicone oils into thin droplets and enhance smoothness and flexibility in fabrics. In water repellents, they support uniform application of protective coatings that lead to persistent and uniform performance. Their resistance to heat is also in line with their use in high temperature finishing treatments. In textile processes, improvement in fabric characteristics and performance traits. As softening aids, ethoxylated fatty amides and silicone-based non-ionic such as polydimethylsiloxane derivatives give outstanding hand feel to fabrics without yellowing effects which are commonly seen in cationic softeners. The rebounded polyglycerol esters which can both give softness and still ensure moisture wicking in sportswear. In lubricant uses, polyethylene glycol derivatives and ethoxylated castor oils minimize fibre-to-fibre and fibre-to-metal friction during mechanical processing, avoiding damage to fragile fibres such as silk and fine wool. Anti-static treatments often employ ethoxylated amines and polyglycol ethers to drain electrical charges on synthetic fibres. Perhaps most importantly, non-ionic surfactants allow for exact control of fabric surface properties via hydrophilic/hydrophobic modification. Fluorinated non-Ionics produce long-lasting water-repelling finishes for outdoor wear, and ethoxylated sorbitan esters yield hydrophilic finishes to make moisture management in performance fabrics better. [43-48]

### 4. Conclusion

Non-ionic surfactants are a backbone in the development of contemporary textile processing, providing multifunctional advantage that caters to both technological and environmental requirements. Due to their distinctive characteristics such as superior chemical compatibility, low toxicity, and wide functionality, they can be applied to all the pre-treatment, dyeing, and finishing phases. Ranging from the enhancement of dye uniformity and fabric softness to providing sustainable processes with less environmental impact, non-ionic surfactants have proved to be unparalleled in their versatility. The development of bio-based and smart surfactants further complements their applications in future-proof textile technologies as the industry becomes more geared towards green chemistry and performance textiles. With tightening environmental regulations and changing consumer needs, non-ionic surfactants will play a crucial role in developing textile formulations that are effective, sustainable, and innovative. Continued R&D will play a vital role in

maximizing their use and revealing new functions, rendering them as vital agents for the revolutionizing of the textile industry.

## 5. REFERENCE

1. H. Cortés et al., “non-ionic surfactants for stabilization of polymeric nanoparticles for biomedical uses,” *materials*, vol. 14, no. 12, p. 3197, jun. 2021, doi: 10.3390/ma14123197.
2. A. Hassabo, b. Hegazy, a. Sediek, f. Saad, and h. Ghazal, “the use of non-ionic surfactants in the textiles industry,” *j. Text. Color. Polym. Sci.*, vol. 0, no. 0, pp. 0–0, jul. 2023, doi: 10.21608/jtcps.2023.216813.1188.
3. M. Sato and a. Quye, “detergency evaluation of non-ionic surfactant dehypon® Is54 for textile conservation wet cleaning,” *j. Inst. Conserv.*, vol. 42, no. 1, pp. 3–17, jan. 2019, doi: 10.1080/19455224.2018.1556719.
4. Husmann, s., munzar, m., & wolfgang, w. (2013). Use of a surfactant composition for the hydrophilic finishing of textile fibers and textile products manufactured therefrom. <https://www.freepatentsonline.com/y2016/0281293.html>
5. N. Saxena, n. Pal, k. Ojha, s. Dey, and a. Mandal, “synthesis, characterization, physical and thermodynamic properties of a novel anionic surfactant derived from sapindus laurifolius,” *rsc adv.*, vol. 8, no. 43, pp. 24485–24499, 2018, doi: 10.1039/c8ra03888k.
6. R. A. Martínez, d. C. Fechner, m. R. Delfino, r. G. Pellerano, and h. C. Goicoechea, “rapid determination of three textile surfactants in environmental samples by modeling excitation-emission second-order data with multi-way calibration methods,” *environ. Sci. Pollut. Res.*, vol. 29, no. 17, pp. 25869–25880, apr. 2022, doi: 10.1007/s11356-021-17660-7.
7. O. E. P. Smith, l. J. Waters, w. Small, and s. Mellor, “cmc determination using isothermal titration calorimetry for five industrially significant non-ionic surfactants,” *colloids surf. B biointerfaces*, vol. 211, p. 112320, mar. 2022, doi: 10.1016/j.colsurfb.2022.112320.
8. A. J. Sayyed, n. A. Deshmukh, and d. V. Pinjari, “a critical review of manufacturing processes used in regenerated cellulosic fibres: viscose, cellulose acetate, cuprammonium, licl/dmac, ionic liquids, and nmmo based lyocell,” *cellulose*, vol. 26, no. 5, pp. 2913–2940, mar. 2019, doi: 10.1007/s10570-019-02318-y.
9. K.-h. Oh, j. R. Baran, w. H. Wade, and v. Weerasooriya, “temperature insensitive microemulsion phase behavior with non-ionic surfactants,” *j. Dispers. Sci. Technol.*, vol. 16, no. 2, pp. 165–188, mar. 1995, doi: 10.1080/01932699508943667.
10. C. Costa, a. Viana, i. S. Oliveira, and e. F. Marques, “interactions between ionic cellulose derivatives recycled from textile wastes and surfactants: interfacial, aggregation and wettability studies,” *molecules*, vol. 28, no. 8, p. 3454, apr. 2023, doi: 10.3390/molecules28083454.
11. Z. A. Yaminzoda (yaminova), “study of the properties of surface-active substances determining the efficiency of dyeing and rinsing of textile materials,” *technol. Qual.*, vol. 55, no. 1, pp. 29–34, apr. 2022, doi: 10.34216/2587-6147-2022-1-55-29-34.
12. W. S. Atikah, i. Muslim, and s. M. Pratama, “the effect of surfactant on pre-mordanting and ph of dyeing process with natural dyes,” *j. Kim. Ris.*, vol. 8, no. 2, pp. 156–166, dec. 2023, doi: 10.20473/jkr.v8i2.49114.
13. B. Chethan, v. Prasad, a. Sunil kumar, s. Thomas, and m. P. Rupashree, “fabrication of surfactant-based electrochemical sensors and biosensors for analysis of textile dyes,” in *surfactant based electrochemical sensors and biosensors*, elsevier, 2024, pp. 163–176. Doi: 10.1016/b978-0-443-15493-5.00003-8.
14. J. Szejtli, “cyclodextrins in the textile industry,” *starch - Stärke*, vol. 55, no. 5, pp. 191–196, may 2003, doi: 10.1002/star.200390050.
15. A. Anceschi, c. Riccardi, and a. Patrucco, “the role of ionic liquids in textile processes: a comprehensive review,” *molecules*, vol. 30, no. 2, p. 353, jan. 2025, doi: 10.3390/molecules30020353.
16. V. Nace, *nonionic surfactants: polyoxyalkylene block copolymers*, 1st ed. Crc press, 2017. Doi: 10.1201/9780203745656.
17. N. M. Van os, ed., *nonionic surfactants: organic chemistry*, 1st ed. Crc press, 2017. Doi: 10.1201/9780203745649.
18. D. Balzer and h. Luders, eds., *nonionic surfactants: alkyl polyglucosides*, 0 ed. Crc press, 2000. Doi: 10.1201/9781482273212.

19. C. Pacifico and s. Giers, "surfactants used in textile applications," *j. Am. Oil chem. Soc.*, vol. 32, no. 4, pp. 231–235, apr. 1955, doi: 10.1007/bf02649020.
20. C. Kim and y.-l. Hsieh, "wetting and absorbency of nonionic surfactant solutions on cotton fabrics," *colloids surf. Physicochem. Eng. Asp.*, vol. 187–188, pp. 385–397, aug. 2001, doi: 10.1016/s0927-7757(01)00653-7.
21. S. M. F. Kabir, s. Chakraborty, s. M. A. Hoque, and k. Mathur, "sustainability assessment of cotton-based textile wet processing," *clean technol.*, vol. 1, no. 1, pp. 232–246, sep. 2019, doi: 10.3390/cleantechnol1010016.
22. C. Carnero ruiz, ed., *sugar-based surfactants: fundamentals and applications*, 0 ed. Crc press, 2008. Doi: 10.1201/9781420051674.
23. J. M. N. Marikkar, h. M. Ghazali, y. B. Che man, t. S. G. Peiris, and o. M. Lai, "distinguishing lard from other animal fats in admixtures of some vegetable oils using liquid chromatographic data coupled with multivariate data analysis," *food chem.*, vol. 91, no. 1, pp. 5–14, jun. 2005, doi: 10.1016/j.foodchem.2004.01.080.
24. C. Obi and m.-a. N. Mgbemena, "review on development and application of eco-benign surfactants," in *surfactants - fundamental concepts and emerging perspectives*, o. Owoseni, ed., intechopen, 2024. Doi: 10.5772/intechopen.110867.
25. G. Buschle-diller, y. El mogahzy, m. K. Inglesby, and s. H. Zeronian, "effects of scouring with enzymes, organic solvents, and caustic soda on the properties of hydrogen peroxide bleached cotton yarn," *text. Res. J.*, vol. 68, no. 12, pp. 920–929, dec. 1998, doi: 10.1177/004051759806801207.
26. Jelinek, c. F., & mayhew, r. L. (1954). *Nonionic surfactants: their chemistry and textile uses*. *Textile research journal*, 24(8), 765-778.
27. Mukaddam akhmedova. (2024). *The role of surfactants in the bleaching process of natural fiber textile products*. *International multidisciplinary journal for research & development*, 11(04).
28. Sk. M. Raafi, s. N. Arju, md. Asaduzzaman, h. H. Khan, and md. Rokonzaman, "eco-friendly scouring of cotton knit fabrics with enzyme and soapnut: an alternative to conventional naoh and synthetic surfactant-based scouring," *heliyon*, vol. 9, no. 4, p. E15236, apr. 2023, doi: 10.1016/j.heliyon.2023.e15236.
29. A. Y. L. Tang and c. Kan, "non-aqueous dyeing of cotton fibre with reactive dyes: a review," *color. Technol.*, vol. 136, no. 3, pp. 214–223, jun. 2020, doi: 10.1111/cote.12459.
30. G. Varadarajan and p. Venkatachalam, "sustainable textile dyeing processes," *environ. Chem. Lett.*, vol. 14, no. 1, pp. 113–122, mar. 2016, doi: 10.1007/s10311-015-0533-3.
31. N. Ristić, "the effect of nonionic surfactant treatment on dyeing of cotton fabrics
32. M. Kert and b. Simončič, "the influence of nonionic surfactant structure on the thermodynamics of anionic dye-cationic surfactant interactions in ternary mixtures, dyes pigments," vol. 79, no. 1, pp. 59–68, oct. 2008, doi: 10.1016/j.dyepig.2008.01.005.
33. Y. L. A. Tang et al., "reverse micellar dyeing of cotton fabric with reactive dye using biodegradable non-ionic surfactant as nanoscale carrier: an optimisation study by one-factor-at-one-time approach," *polymers*, vol. 15, no. 20, p. 4175, oct. 2023, doi: 10.3390/polym15204175.
34. A. Y. L. Tang, c. H. Lee, y. Wang, and c.-w. Kan, "polyethylene glycol (peg) non-ionic surfactant-based reverse micellar dyeing of cotton fabric with hot type trichloropyrimidine (tcp)-based reactive dyes," *j. Nat. Fibers*, vol. 20, no. 1, p. 2131689, apr. 2023, doi: 10.1080/15440478.2022.2131689.
35. S. Md. Mamun kabir and j. Koh, "dyeing chemicals," in *chemistry and technology of natural and synthetic dyes and pigments*, a. Kumar samanta, n. S. Awwad, and h. Majdooa algarni, eds., intechopen, 2020. Doi: 10.5772/intechopen.81438.
36. Y. L. A. Tang, c. H. Lee, c. Y. Chan, y. Wang, and c.-w. Kan, "alkyl polyglucoside (apg) nonionic surfactant-based reverse micellar dyeing of cotton fabric – a study of reactive dyes with different functional groups," *j. Nat. Fibers*, vol. 20, no. 1, p. 2136324, apr. 2023, doi: 10.1080/15440478.2022.2136324.
37. K. Yun and c. Ahn, "effect of surfactant type on the dyeability and color resistance of semi-permanent basic hair dye," *fash. Text.*, vol. 10, no. 1, p. 4, jan. 2023, doi: 10.1186/s40691-022-00326-4.
38. L. F. Guimarães and m. A. Granato, "dyeing of polyamide fabrics with a reverse micellar system using soybean oil as a solvent," *color. Technol.*, vol. 140, no. 1, pp. 103–113, feb. 2024, doi: 10.1111/cote.12702.
39. Y. Nemoto and h. Funahashi, "the interaction between dyes and nonionic surfactants: the mode of action on nonionic surfactants in dyeing," *ind. Eng. Chem. Prod. Res. Dev.*, vol. 19, no. 2, pp. 136–142, jun. 1980, doi: 10.1021/i360074a003.

40. Werner I. The mode of action of non-ionic levelling agents. J.s.d.c. 1958;74(4):221-233. Issn: 1478-4408
41. Valko ei. Textile auxiliaries in dyeing. Review progress in coloration. 1972;3(1):50-62. Issn: 1478-4408
42. jones f. The theory of dyeing. Review progress in coloration. 1967;1(1):15-22. Issn: 1478-4408
43. C.n. sivaramakrishnan, the use of surfactants in the finishing of technical textiles, in the dyeing and finishing of technical textiles, woodhead publishing,2013, pages 199-235, isbn 9780857094339, <https://doi.org/10.1533/9780857097613.2.199>.
44. iheaturu, nnamdi & aharanwa, bibiana & chike, kate & ezeamaku, engr. Dr uchenna & nnorom, onyekachi & chima, chibueze & nnorom, onyekachi. (2019). Advancements in textile finishing. 6. 23-31. 10.9790/019x-06052331.
45. Gulrajani, m. L., & gupta, d. (2011). Emerging techniques for functional finishing of textiles. Indian journal of fibre & textileresearch, 36(december), 388–397. *(pdf) advancements in textile finishing.*
46. Joshi, m., & butola, b. S. (2013). Application technologies for coating, lamination and finishing of technical textiles. In m. L.gulrajani (ed.), advances in the dyeing and finishing of technical textiles (pp. 355–411). Woodhead publishing. Retrieved from <http://www.sciencedirect.com/science/article/pii/B9780857094339500146> *(pdf) advancements in textile finishing.*
47. Yang, s. L. L. (1991). Amino-functional siloxanes and hydrolyzable silane for finishing textiles. Eur. Pat. Appl. Union carbide chemicals and plastics co., inc., usa . *(pdf) advancements in textile finishing.*
48. Madhusmita tarai, saroj yadav "advances in textile finishing", international journal of science & engineering development research ([www.ijrti.org](http://www.ijrti.org)), issn:2455-2631, vol.9, issue 4, page no.26 - 32, april-2024